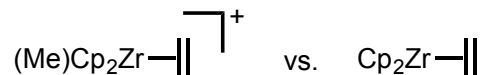


TODAY'S TOPICS

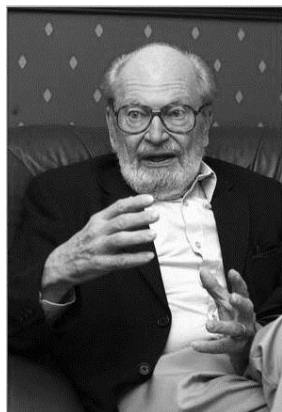
- *trans* effect/influence
- ligand exchange
- oxidative addition

PROBLEMS OF THE DAY

- #1** Consider the following two complexes and the ability of the metal in both cases to participate in back-bonding. **Provide an explanation for why one of the species results in a more stable complex than the other.**

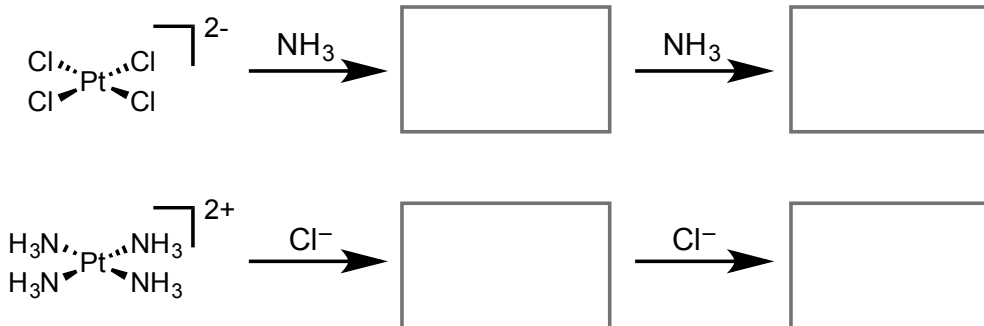


CHEMIST OF THE DAY



name?
known for?

- #2** On the basis of the *trans* effect, **predict the products in the following reaction sequences:**

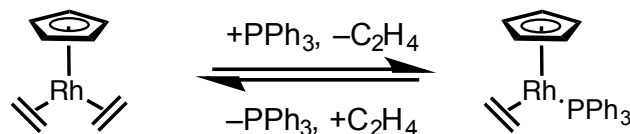


QUOTE OF THE DAY

"Ever tried. Ever failed. No matter. Try again. Fail again. Fail better."

- Samuel Beckett

- #3** Consider the following ligand exchange process:



A. Give the coordination number, oxidation state, and total electron count for both complexes.

B. Design one or more experiments to determine the mechanism of ligand exchange (associative vs. dissociative).

READING

Hartwig: Ch. 5–7
Crabtree: Ch. 1.4, 4.4–4.7,
6.1–6.5