

TODAY'S TOPICS

- organometallic history
- electron counting
- 18-electron rule
- structure and bonding

CHEMIST OF THE DAY



QUOTE OF THE DAY

"The real voyage of discovery consists not in seeking new landscapes, but in having new eyes."

- Marcel Proust

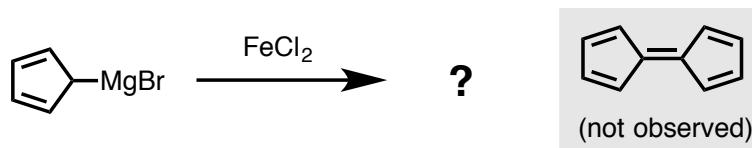
READING

Hartwig: Ch. 1.1–1.3
Crabtree: Ch. 1–2

PROBLEMS OF THE DAY

#1

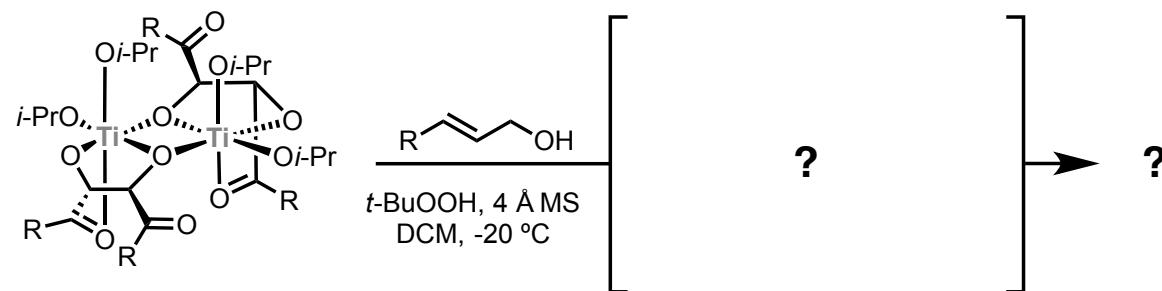
A. In 1951, Paulson and Kelly attempted to synthesize fulvalene under the conditions shown below. Instead they obtained an unexpected product. **Provide as many possible connectivities as possible for this unexpected product and propose how to rule out alternatives using techniques available in the 1950s.**



B. For the correct structure, **provide the coordination number, metal oxidation state, d-electron count, and overall electron count.**

#2

A. The Sharpless titanium-tartrate catalyst exists as a bridged dimer. **Provide the coordination number, metal oxidation state, d-electron count, and overall electron count.**



B. **Provide the proposed dimeric intermediate that leads to product formation and predict the stereochemical outcome.** (Hint: Coordination number and oxidation state remain unchanged.)

#3

For the structures below, **provide the coordination number, metal oxidation state, d-electron count, and overall electron count.** Bonus: **provide the names of these catalysts.**

